Polycyclic Aromatic Hydrocarbons Pollution in a Coastal Environment: the Statistical Analysis of Dependence to Estimate the Source of Pollution

Pasquale Sarnacchiaro¹, Sergi. Díez² and Paolo Montuori^{3,*}

¹Unitelma Sapienza University of Rome, V.le Regina Elena nº 295, Roma 00161, Italy

²Environmental Chemistry Department, IDAEA-CSIC, Jordi Girona, 18–26, 08034 Barcelona, Spain

³Department of Preventive Medical Sciences, University of Naples "Federico II", Via Sergio Pansini n^o 5, Naples 80131, Italy

Abstract: Polycyclic Aromatic Hydrocarbons (PAHs) are a group of carcinogenic contaminants widespread in the environment. PAHs are produced by both anthropogenic and natural processes. Difficulties exist in identifying their origins. This paper reports a practical application of Principal Component Analysis (PCA) and Principal Component Regression (PCR) to identify the pyrolytic, petrogenic and diagenesis sources of PAH pollution in the Sarno River and Estuary. Nick-named "the most polluted river in Europe", the Sarno River originates in south-western Italy and has a watershed of about 715 km². PCA indicated that the PAH contamination in the Sarno River and Estuary resulted from a mixed pattern. The first principal component (PC1) had significant positive loading in high molecular weight PAHs. This profile of PAH usually includes products of high temperature combustion/pyrolitic processes, reflecting the effects of traffic pyrolysis. The second principal component (PC2) had significant positive loading in two-to-four ring PAHs. So, PC2 may be considered as components from petrogenic sources. PC3 was characterized by a high loading of perylene, thought to originate from diagenetic alteration of perylenequinone pigment or some other organic matter. Therefore, this factor can be considered as natural-origin PAHs. In the PCR, the regression coefficients for components 1-3 were 66.6, 40.4 and 19.5, respectively. In this application, the PCR was a very useful statistical technique for handling the problem of multicollinearity. Results from the application of PCR have been compared with Partial Least Square (PLS) and no significant differences were reported in the prediction errors and latent variables available by PCR and PLS.

Keywords: Polycyclic aromatic hydrocarbons, Coastal environment, Contaminant transport processes, Principal component analysis, Principal Component Regression, Multicollinearity.

INTRODUCTION

Polycyclic aromatic hydrocarbons (PAHs) are a group of lipophilic contaminants widespread in the environment. This class of compounds has been widely studied [1, 2] because of its carcinogenic and mutagenic properties [3-5].

PAHs are produced by both anthropogenic and natural processes and can be introduced into the environment through various routes. Anthropogenic inputs can originate from the incomplete combustion of organic matter (pyrolytic) and the discharge of crude oil-related material (petrogenic). The pyrogenic origin includes domestic and industrial wastewater discharges as well as atmospheric fallout of vehicle exhaust and industrial stack emission, combustion of fossil fuels, waste incineration and coke production, carbon black, coal tar pitch, asphalt and petroleum cracking [6]. Spillage of fossil fuels can be accidental or intentional (e.g. deballasting water discharges). PAHs can also originate from natural processes such as short-term diagenetic degradation of biogenic precursors (diagenesis). Direct PAH biosynthesis

by organisms such as bacteria, fungus and algae has not yet been clarified [7]. Each source (i.e. pyrolytic, petrogenic and diagenetic) gives rise to characteristic PAH patterns. In general, combustion products are dominated by relatively high molecular weight (HMW) compounds with four or more condensed aromatic rings, whereas bi- and tricyclic aromatic compounds (LMW) are more abundant in fossil fuels, which are, moreover, dominated by alkylated derivatives [8]. Diagenetic PAHs are represented by some characteristic species (e.g. perylene). Using chemical profiling and particular compound ratios, it is possible can get an idea of the process that generates these compounds [9-11]. The ratios of phenanthrene to anthracene (Phe/An) and fluoranthene to pyrene (Fl/Pyr) have been used to assess the contribution of petroleum or combustion sources of PAHs in coastal environments. A Phe/An ratio lower than 10 and a Fl/Pyr ratio higher than 1 have been shown in the majority of cases, strongly supporting the pyrogenic origin of PAHs [12]. In addition to this, when the ratio of chrysene to benzo(a)anthracene (Chr/BaA) is lower than 0.1, it is usually an indication of a pyrolitic origin, while a fluorene to fluorene + pyrene (Flu/(Flu + Pyr)) ratio lower than 0.5 is typical of a petrogenic source. However, difficulties exist in identifying these origins owing to the possible co-existence of several variables related to different sources.

^{*}Address correspondence to this author at the Department of Preventive Medical Sciences, University of Naples "Federico II", Via Sergio Pansini nº 5, Naples 80131, Italy; Tel: +390817463027; Fax: +39081746 3352; E-mail: pmontuor@unina.it

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Currently, the interest in multivariate statistical methodologies for identifying the main sources of PAH pollution and for quantifying the incidence of each source of pollution on total pollution levels, particularly in coastal environments, is increasing [13-15]. Diagnostic ratios provide only qualitative information on the contributions of various sources to total PAH levels. Quantitative assessments can be done with several statistical approaches such as Principal Component Analysis (PCA) and Multiple Linear Regression Analysis (MLRA) [14, 15].

In many frameworks, the statistician is interested in the relationship between a set of variables, called predictors, and one response variable, both defined for the same statistical units. In regression, the objective is to explain the variation in one response variable by associating this variation with proportional variation in one or more predictor variables. When strong correlations exist between the predictor variables, the ordinary least squares estimators have large variances and, therefore, the estimates are often large and may have signs that disagree with known theoretical properties of the variables. In the statistical literature, several methods have been proposed to counter the multicollinearity problem [20], for example Ridge Regression (RR), Principal Component Regression (PCR), Partial Least Square Regression (PLS) and Principal Component Analysis onto References subspaces (PCAR) [21]. PCR shows moderate similarity with the other methods, probably because it transforms the predictor variables on the orthonormal basis of latent variables by eliminating the collinearity between predictor variables [22].

PCR has been introduced to handle the problem of multicollinearity by eliminating model instability and reducing the variances of the regression coefficients [23]. In the first step, the methodology performs a PCA on the predictor variables, and in the second step runs a regression using the principal component scores as the predictor variables with the response variable of interest. A first question in PCR is which principal component should be kept in the regression. Many statisticians argue that the decision depends only on the magnitude λ of the variance of the principal components [24]; however, there are some examples in which the principal components of small variance must be selected [25]. The main problem of this approach is that the magnitude of the eigenvalue (λ) depends on the predictor variables only and has nothing to do with the response variable; in other words, there is no general theoretical reason that the most informative linear function of the predictor variable should lie among the dominant principal components of the multivariate distribution of the predictor variables. However, when the interpretation of Principal Components is clear and useful, PCR represents a good solution to overcome the multicollinearity problem [26, 27].

This study is part of a large project which has the objective of enhancing knowledge of pollution in the Sarno River and its environmental impact in the gulf of Naples. This project has attempted to assess the pollution derived from local industry, agriculture and urban impact through the determination of several groups of organic and inorganic chemicals and some indicators of microbial pollution in water and sediments. In the present work, we report a practical application of PCA and PCR to identify the pyrolytic, petrogenic and diagenesis origin of PAH pollution in sediments of the Sarno River and Estuary.

EXPERIMENTAL

Study Area

Nicknamed "the most polluted river in Europe", the Sarno River originates in south-western Italy and has a watershed of about 715 km². It flows through the Sarno flatland, delimited in the west by Mt. Vesuvius and in the east by the Lattari Mountains, and reaches the Sea at the Gulf of Naples (Tyrrhenian Sea), flowing through Pompei city (Fig. 1).

The Sarno watershed collects water from two important effluents, the Cavaiola and Solofrana torrents. The Sarno flatland is one of the most fertile in Italy due to the high quality of the soil, constituted by layers with volcanic and alluvial origins. The high human presence, the massive use of fertilisers and pesticides in agriculture and the development of industry represent the main causes of pollution in the Sarno River [16]. Agriculture is focused on tomato production in the San Marzano area. In terms of industry, Solofra, a city around the Solofrana River, boasts an ancient tradition in the leather industry that currently counts for about 400 productive units and 3500 workers. The pharmaceutical industry is represented principally by the Novartis Farma factory, placed exactly 200 m from the river mouth and covering an area of about 201,000 m^2 . This factory is one of the largest facilities of Novartis Farma and is one of the most important in the world. Another source of environmental pollution can be attributed to urban agglomerations and their wastewaters [17-19].

Sampling Points and Sediment Sample Collection

An intensive sampling campaign was conducted in the spring of 2008 (Fig. 1). Surface sediment samples were collected at four locations along the Sarno river (near the source of the river, just before and after the junction with Alveo Comune and at the river mouth) and nine points in the continental shelf around the river mouth (three points were sampled 50 m from the Sarno River mouth, another three points 150 m away and, finally, another three points 500 m from the river mouth).

Surface sediment (0-20 cm) samples were collected by using a grab sampler (Van Veen Bodemhappe 2 L) and put in aluminium containers. The sediments were transported refrigerated to the laboratory and kept at -20°C before analysis.

Sediment Sample Extraction

Sediments were oven dried at 60°C and sieved at 250 μ m. Then, 5 g of sediment were spiked with the surrogate mixture (10 ng of anthracene- d_{10} , pyrene- d_{10} and peylene- d_{12}) and extracted three times by sonication using 15 mL of DCM/methanol (1:1) (Carlo Erba, Milano, Italy) for 15 min. The recovered extracts were combined and dried with anhydrous sodium sulphate (Carlo Erba). Then, they were concentrated in a rotary evaporator to 0.5 mL and solventexchanged to hexane (Carlo Erba). Extract fractionation was



Fig. (1). Map of the study areas and sampling sites in the Sarno River and Estuary, Southern Italy (from Google Hearth).

carried out by open column chromatography (3 g of neutral alumina Carlo Erba, deactivated with 3% (w/w) Milli-Q water). Three fractions were collected: fraction I with 5.5 mL of hexane (Carlo Erba), fraction II with 6 mL of hexane:ethylacetate (9:1) (Carlo Erba) and, finally, fraction III with 12 mL of ethylacetate (fraction III or F3). PAHs were eluted in fraction II, while fractions I and III contained other organic pollutants which were also determined in the samples.

Analytical Determination of PAHs

Cleaned extracts of fractions II were analysed on a GC-MS QP5050A Shimadzu (Kyoto, Japan) working in the electron impact mode at 70 eV. A SPB 20 chromatographic column (20% diphenyl 80% dimethylpolysiloxane) (60 m, 0.25 mm ID and 0.25 μ m of film thickness) was used. The initial oven temperature was temperature programmed at 50°C (2 min) to 250°C at 10°C min⁻¹, holding for 30 min. The MS transfer line and ion source were kept at 250°C. Acquisition was carried out in the single ion monitoring mode (SIM) using two characteristic ions for each target analyte. Compound identification was carried out by comparing retention times with standards and using the characteristic ions and their ratio for each target analyte. Furthermore, for the higher concentrated samples, the identification of target analytes was confirmed in full-scan mode (m/z range from 60 to 350). The concentrations were calculated from the calibration curves for the 17 PAHs, and a good chromatographic linearity ($\mathbb{R}^2 > 0.98$) of the Standard Reference Material (Dr. Ehrenstorfer GmbH, Augsburg, Germany) was found. Triphenylamine was used as the internal standard to compensate for the sensitivity variation of the MS detector. In each sample, the concentration of the following sixteen selected PAHs monitored by the US Environmental Protection Agency (USEPA) as priority pollutants and perylene were measured: naphthalene (Nap), acenaphthylene (Acy), acenaphtalene (Ace), fluorene (Flu), phenanthrene (Phe), anthracene (An), fluoranthene (F1), pyrene (Pyr), benz(a)anthracene (BaA), chrysene (Chr), benzo(b) fluoranthene (BbF), benzo(k)fluoranthene (BkF), benzo(a)pyrene (BaP), dibenz(a,h)anthracene (DahA), benzo(ghi)perylene (BghiP), indeno (1,2,3-cd) pyrene (InD) and perylene. Total PAH concentrations were calculated as the sum of the concentrations of the 16 PAH compounds (Σ PAHs) (Table 1).

Quality Assurance and Quality Control

The limit of detection (LOD) and limit of quantification (LOQ) were calculated as having signal-to-noise ratios of

	Sampling Location					
Site Number	Site	Site	(ng g ⁻¹ dry wt)			
Identification	Characteristics	Location				
l (river water)	Sarno River Source	40°48′54.03″N 14°36′45.36″E	5.6			
2	Upstream	40°46′42.73″N	101.3			
(river water)	Alveo Comune	14°34′00.48″E				
3	After	40°46′00.34″N	205.2			
(river water)	Alveo Comune	14°33′10.68″E				
4	Sarno River	40°43'42.62"N	352.4			
(river water)	Mouth	14°28'07.89"E				
5	River Mouth	40°43'40.11"N	446.6			
(sea water)	at 50mt South	14°28'06.45"E				
6	River Mouth	40°43'42.46"N	501.7			
(sea water)	at 50mt Central	14°28'05.03"E				
7	River Mouth	40°43′45.09″N	180.4			
(sea water)	at 50mt North	14°28′05.17″E				
8	River Mouth	40°43′35.68″N	651.3			
(sea water)	at 150mt South	14°28′02.94″E				
9	River Mouth	40°43′42.25″N	679.2			
(sea water)	at 150mt Central	14°27′59.97″E				
10	River Mouth	40°43′49.26″N	161.9			
(sea water)	at 150mt North	14°27′59.82″E				
11	River Mouth	40°43′30.31″N	66.5			
(sea water)	at 500mt South	14°27′58.93″E				
12	River Mouth	40°43′42.29″N	87.3			
(sea water)	at 500mt Central	14°27′46.41″E				
13	River Mouth	40°43′57.85″N	40.4			
(sea water)	at 500mt North	14°27′48.68″E				

Table 1. Description of the	Sampling Sites and	Concentration o	of PAHs in the	Sediments of th	ne Sarno Ri	iver and the	Continental
Shelf, Southern Ita	aly						

above 3 and 10, respectively, by five replicate analyses. The surrogate averaged recoveries in the sediment samples were $84.8 \pm 7.8\%$ for anthracene- d_{10} , $91.1 \pm 8.5\%$ for pyrene- d_{10} and $102 \pm 12\%$ for perylene- d_{12} . Blank assays were carried out and used for the calculation of LODs (ranged from 0.01 ng g⁻¹ to 0.15 ng g⁻¹) and LOQs (ranged from 0.03 ng g⁻¹ to 0.2 ng g⁻¹).

Statistical Analysis

Data analysis was performed with the statistical software SPSS, version 14.01 for Windows (SPSS Inc., Chicago, IL, USA) and SPAD, version 5.5 for Windows (Decisia, Pantin, France).

PCA was conducted to identify the source contributions to total PAH concentrations. PCA is a useful exploratory multidimensional data analysis technique for examining factors that contribute to data variability. Data submitted to the analysis were arranged in a matrix, where each column corresponds to one PAH component and each row represents one sample site. The data matrix was evaluated through PCA, allowing the summarised data to be further analysed and plotted.

The statistical analysis of dependence was performed in order to determine the percentage contributions of different PAH sources to the \sum PAH for a given sediment sample. First, to analyse the dependence between the total PAH level and the seventeen selected PAHs, a preliminary study of the correlation between the predictor variables was performed (Table 2). As can be seen in Table 2, strong positive correlations were found between several pairs of predictors. When strong correlations exist between the predictor variables, it induces an inflation of the variance of the least squares estimates, increase in the magnitude of these estimates and leads to unreliable signs of the regression coefficients.

In order to detect collinearity, the variance inflation factor (VIF) were computed. VIF resulted upper than 10 for all the predictor variables, therefore a serious collinearity problems were observed. A statistical collinearity between the

Correlation Matrix	Nap	Acy	Ace	Flu	Phe	An	Pyr	Chr	BaA	BbF	BkF	BaP	InD	DahA	BghiP	Fl	Perylene
Nap	1,00																
Acy	0,88*	1,00															
Ace	0,94*	0,77*	1,00														
Flu	0,85*	0,93*	0,73*	1,00													
Phe	0,88*	0,88*	0,81*	0,79*	1,00												
An	0,92*	0,93*	0,84*	0,94*	0,82*	1,00											
Pyr	0,84*	0,98*	0,72*	0,96*	0,85*	0,90*	1,00										
Chr	0,90*	0,93*	0,79*	0,83*	0,97*	0,87*	0,89*	1,00									
BaA	0,30	0,21	0,23	0,07	0,05	0,29	0,10	0,22	1,00								
BbF	0,27	0,16	0,18	0,03	0,00	0,23	0,06	0,18	0,99*	1,00							
BkF	0,27	0,18	0,15	0,03	0,06	0,24	0,08	0,23	0,98*	0,98*	1,00						
BaP	0,27	0,17	0,20	0,06	-0,02	0,26	0,07	0,15	0,99*	0,99*	0,97*	1,00					
InD	0,24	0,13	0,16	0,01	-0,02	0,22	0,03	0,16	0,99*	0,99*	0,98*	0,99*	1,00				
DahA	0,22	0,11	0,14	-0,02	-0,04	0,19	0,01	0,14	0,99*	1,00*	0,98*	0,99*	1,00*	1,00			
BghiP	0,14	0,04	0,08	-0,08	-0,12	0,13	-0,07	0,07	0,97*	0,98*	0,96*	0,98*	0,99*	0,99*	1,00		
Fl	0,29	0,19	0,21	0,06	0,04	0,27	0,09	0,21	1,00*	1,00*	0,99*	0,99*	1,00*	1,00*	0,98*	1,00	
Perylene	-0,23	-0,34	-0,40*	-0,29	-0,38	-0,18	-0,37	-0,27	0,40*	0,40*	0,47*	0,14	0,47*	0,45*	0,48*	0,42*	1,00

p-value < 0.01

independent variables can be also detected by the proximity of the smallest eigenvalues of the correlation matrix R to 0 (Table 4).

To overcome multicollinearity problems, a large number of statistical methods are available: RR, PCR, PLS and PCAR. Given the clear popularity of PLS and PCR among chemists we have compared them. These two techniques are similar in many ways and the theoretical relationship between them has been treated extensively in the literature [28]. Both methods construct new predictor variables, known as components, as linear combinations of the original predictor variables, but they construct those components in different ways. PCR creates components to explain the observed variability in the predictor variables, without considering the response variable at all. On the other hand, PLS does take the response variable into account, and therefore often leads to models that are able to fit the response variable with fewer components. Whether or not that ultimately translates into a more parsimonious model, in terms of its practical use, depends on the context.

Historically, PCR predates PLS, with the latter appearing in the chemical literature around 1983 [28]. Since its introduction, however, PLS appears by most accounts to have become the method of choice among chemists. The main reasons are: (1) PLS should predict better because correlations with the y variable are sought in determining the scores, (2) PLS requires fewer latent variables than PCR and should therefore be more parsimonious, (3) PLS loadings are more readily interpreted, and (4) PLS should handle nonlinearities better than PCR.

In their research P.D. Wentzell and L.V. Montoto considered many papers in order to obtain a clear picture of what kinds of studies had already been done to compare these two methods and what conclusions had been drawn. Their conclusion was "The results of the study were somewhat surprising. Given the clear popularity of PLS over PCR among chemists, we had expected to see many instances where the superior predictive performance of PLS was evident. While there were a few cases which indicated that PLS gave better results than PCR, a greater number of studies indicated no real difference in performance. Additionally, there were no theoretical studies which suggested that one method should predict better than the other"(pag.3).

Also in our research, results from the application of PCR, compared with PLS, shown no significant differences in the prediction errors and latent variables available.

PCR was preferred since the interpretation of the extracted principal components had a clear reading in relation to research, in particular the first principal component had significant positive loading in high molecular weight PAHs and this profile reflects the effects of traffic pyrolysis. The second principal component had significant positive loading in two-to-four ring PAHs, then it may be considered as components from petrogenic sources. The third component was

Table 3. Concentration	Ranges and Mean	Value of PAHs in	Sediment Samples	from Different	Rivers, Estuaries an	d Coasts in the
World						

Steriler Arrise	Deferment	N. of PAHs	Sediments (ng G ⁻¹)		
Study Area	References	Analyzed	Range	Mean	
Asia					
Dalaio River watershed, China	Guo et al. (2009)	18	62 - 841	287	
Qiantang River, China	Chen et al. (2007)	15	91 - 614	313	
Lanzhou Reach, Yellow River, China	Xu et al. (2007)	16	464 - 2621	1414	
Tonghui River, China	Zhang et al. (2004)	16	127 - 928	540	
Kishon River, Israel	Oren et al. (2006)	22	59 - 298	153	
Pearl River Estuary, China	Luo <i>et al</i> (2006)	17	189 - 637		
Australia/Oceania					
Sydney Harbour, Australia	McCready (2000)	16	<0.1 - 380.2		
North America					
Susquehanna River, USA	Ko et al. (2007)	36		3000	
Athabasca River, Canada	Headley et al. (2001)	16	10 - 34700		
Todos Santos Bay, Mexico	Macias et al. (2002)	16	7 - 813	96	
Mediterranean sea					
Adriatic Sea	Guzzella et al.(1994)	9	27 - 530	130	
Adriatic Sea	Caricchia (1993)	12	18 - 580	200	
Gulf of Rijeka, Croatia	Bihari <i>et al.</i> (2006)	10	213 - 695		
Izmir Bay, Turkey	Darilmaz (2007)	16	2 - 113	24	
Tiber River, Italy	Minissi et al. (1998)	13	4 - 625		
Adriatic Sea (Chioggia), Italy	Magi et al. (2002)	16	24 - 501	333	
Adriatic Sea (Ancona), Italy	Magi et al. (2002)	16	33 - 306	184	
Gulf of Trieste, Italy	Notar <i>et al.</i> (2001)	22	30 - 600		
This study		17	5.6-679.2	267.7	

characterized by a high loading of perylene and it can be considered as natural-origin PAHs.

Criteria for the selection of principal components in regression are: Student's tests of the significance of the components or the magnitude of the eigenvalues of the explanatory variables correlation matrix. The two procedures are not quite compatible and a third procedure based on a decision rule with respect to the partial correlation coefficient between the principal component and the explained variable has been proposed [29]. In this research we considered three components.

Moreover, in this study, the concentrations of 17 PAHs as active variables and 13 sediment samples as subjects were selected.

RESULTS AND DISCUSSION

The concentrations of total PAHs in sediment samples are illustrated in Table 1. In sediment samples the mean and the standard deviation of total PAHs concentrations were 267.7 and 233.4 ng g⁻¹, respectively. The detected concentrations ranged from 0.2 to 31.6 ng g⁻¹ with a mean value of 9.7 ng g⁻¹ for two-ring PAHs (Nap), from 0.2 to 46.3 ng g⁻¹ for three-ring PAHs (Acy, Ace, Flu, Phe, An), from 0.3 to 47.2 ng g⁻¹ for four-ring PAHs (Fl, Pyr, BaA, Chr), from 0.2 to 46.6 ng g⁻¹ for five-ring PAHs (BbF, BkF, BaP, DahA) and from 0.5 to 46.7 ng g⁻¹ for six-ring PAHs (BghiP, InD).

Compared with other polluted rivers in the world (Table 3), the concentrations of \sum PAHs in sediment samples from the Sarno River (5.6 - 352.4 ng g⁻¹) were close to those found in the Daliao River watershed (China) by Guo *et al.* (2009) and in the Qiantang River (China) by Chen *et al.* (2007) [28,29]. Values slightly higher than those found in the Sarno River were found in the Tiber River (Italy) by Minissi *et al.* (1998) and in the Tonghui River (China) by Zhang *et al.* (2004) [32, 33], while much higher concentrations were encountered in the Lanzhou Reach of the Yellow River (China) by Xu *et al.* (2007), in the Susqueahanna River (USA) by Ko *et al.* (2007) and in the Athabasca River (Canada) by Headley *et al.* (2001) [15, 34, 35]. The concentrations of \sum PAHs

	T٤	able 4	. Princ	cipal Con	iponent Anal	vsis – Eigen	Values Table
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Eigenvalues Table						
Principal Component	Eigenvalue	Percentage	Percentage Cumulative			
1	8,6968	51,16%	51,16%			
2	6,7744	39,85%	91,01%			
3	0,6997	4,12%	95,12%			
4	0,3997	2,35%	97,47%			
5	0,3124	1,84%	99,31%			
6	0,0542	0,32%	99,63%			
7	0,0317	0,19%	99,82%			
8	0,0218	0,13%	99,95%			
9	0,0052	0,03%	99,98%			
10	0,0031	0,02%	99,99%			
11	0,0008	0,00%	100,00%			
12	0,0002	0,00%	100,00%			
13	0,0000	0,00%	100,00%			
14	0,0000	0,00%	100,00%			
15	0,0000	0,00%	100,00%			
16	0,0000	0,00%	100,00%			
17	0,0000	0,00%	100,00%			

in sediment samples from the Sarno River were higher than those found in the Heman Reach of the Yellow River, China (2009) and the Kishon River, Israel (2006) [36, 37]. **PAHs** in sediments samples from the continental shelf of the Sarno River $(40.3 - 679.2 \text{ ng g}^{-1})$ were close to those found in the Pearl River Estuary (China) by Luo et al. (2006), in the Gulf of Rijeka, Croatia (Bihari et al., 2006) and in the Gulf of Trieste, Italy (Notar et al., 2001) [13, 14, 38]. The concentrations of \sum PAHs in sediment samples from the continental shelf of the Sarno River were higher than those found in Sydney Harbour, Australia (2000), in Todos Santos Bay, Mexico (2002), in Izmir Bay, Turkey (Darilmaz et al., 2007) and in the Adriatic Sea, Italy (1993, 1943 and 2002) [2, 6, 39-42]. The concentrations of Σ PAHs in sediment samples from the continental shelf of the Sarno River were lower than those found in the Venice Lagoon by Manodori et al. (2006) [43].

The ratio study reflected a prevailing pattern of pyrolytic inputs of PAHs in the Sarno River and Estuary [8-12]. In fact, the results showed that the Phe/An ratio was <10 at all sites (mean 1.19; range 0.43-3.62), and Flu/Pyr was >1 at most sites (mean 1.33; range 0.74-4.01). Chr/BaA and Flu/Flu + Pyr varied within a range of 0.12 - 2.07 (mean 0.88) and 0.34 - 1.44 (mean 0.72), respectively. Thus, the LMW/HMW ratio was relatively low (<1 for most sites), suggesting a pyrolitic origin of PAHs at these sites (mean 0.90; range 0.24-3.99).

In addition to pyrolytic and petrogenic sources, perylene is also produced by in situ degradation of biogenic precursors [12, 44]. Indeed, perylene is probably the most important diagenetic PAH encountered in sedimentary environments and, thus, a high abundance of perylene relative to other PAHs can indicate an important natural origin of the compound. The highest concentrations of perylene were observed at location 3 (after Alveo Comune) and between the River and the Estuary. The clearly seaward-decreasing trends of concentrations of perylene suggest that perylene originated mainly from terrestrial inputs. Perylene has been frequently associated with inputs from rivers and estuaries [12, 45]. These authors have suggested that concentrations of perylene which are higher than 10% of the total pentaaromatic isomers indicate a probable diagenetic input, whereas those in which perylene comprises less than 10% indicate a probable pyrolytic origin of the compound. In the present study, perylene was studied in all sediment samples, and concentrations of this compound ranged from 0.12 to 30.3 ng g^{-1} . The concentrations of perylene relative to the penta-aromatic isomers indicated that the relative abundance of some of them were lower than 10%, indicating a pyrolytic origin of the compounds that could reach the river from different sources in this area.

PCA, as mentioned earlier, was used to separate PAHs that had similar sources and modes of input. The majority of the variance (95%) of the scaled data was explained by three eigenvectors-principal components (Table 4).

The first principal component (PC1) explained 51.2% of the total variance, and the second (PC2) and the third (PC3) represented 39.9% and 4.1% of the total variance, respec-

Variable Coordinates	Axes 1	Axes 2	Axes 3
Nap	0,64	0,72	0,00
Acy	0,56	0,80	-0,05
Ace	0,53	0,70	0,24
Flu	0,43	0,83	-0,19
Phe	0,41	0,85	-0,01
An	0,62	0,73	-0,15
Fl	0,61	0,77	0,05
Pyr	0,46	0,84	-0,10
Chr	0,57	0,76	-0,07
BaA	0,92	-0,39	0,06
BbF	0,90	-0,43	0,09
BkF	0,90	-0,41	-0,02
BaP	0,90	-0,42	0,07
InD	0,89	-0,45	0,02
DahA	0,88	-0,47	0,06
BghiP	0,84	-0,54	0,05
Perylene	0,24	-0,60	-0,74

 Table 5. Principal Component Analysis – Variable Coordinates

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Chr Fl Nap 0.8 0.4 0 Ra/ -0.4 Bkl InD BaP ervlene DahA -0.8 -0.4 -0.8 04 0.8

Fig. (2). Principal Component Analysis – Variables Plot (Axes 1-2).



tively. The coordinates for variables show the loadings for the individual PAHs (Table **5**).

Along the axis PC1, all the compounds were found on the positive-coordinate side and none on the negative-coordinate side (Fig. 2), therefore it represents a "size component". Nevertheless, the variables which had the highest influence in building this axes are BaA, BbF, BkF, BaP, InD, DahA and BghiP. These compounds gave strong correlations with the total PAH concentration (r > 0.7). Moreover, it had significant positive loading in high molecular weight PAHs. This profile of PAHs usually implies products of high temperature combustion/pyrolitic processes, reflecting the effects of traffic pyrolysis. PC2 had significant positive loading in two-to-four ring PAHs (Ace, Acy, Flu, An, Fl, Pyr, Phe, Chr and BkF). So, PC2 may be considered a component of petrogenic sources. This also confirms the results obtained from the ratios of specific PAH compounds in that the PAH contamination in the sediments resulted mainly from a mixed pattern of pyrolytic and petrogenic sources.

PC3 was characterised by a high loading of perylene (Fig. 3), thought to originate from diagenetic alteration of perylenequinone pigment or some other organic matter. Therefore, this factor can be considered as natural-origin PAHs.

The PCR results are presented in Table 6. The regression coefficients for Principal Components 1-3 were 66.6, 40.4 and 19.5, respectively. Thus, a mean percentage contribution of 52.6% was from the combustion of coal and wood, 31.9% from oil-derived sources, 25% from vehicle emissions, and

3).

15.4% from natural sources. Compared with the analysis by diagnostic ratios, the results of PCA and PCR are consistent with those of diagnostic ratios. This suggests that both diagnostic ratios and PCA and PCR are appropriate for PAH source identification.

These results obtained by different molecular ratios, PCA and PCR were correlated with the specific pollution conditions in the Sarno flatland. The Sarno flatland is a heavily industrialised area, with many heavily polluting factories. Industrial wastes enriched with combustion-derived PAHs are discharged directly into the Sarno River. The emission of atmospheric particles from factories and from the entire Naples urban area also cause serious air pollution, and particulate-associated PAHs may be transported and deposited

 Table 6. Principal Component Regression – Coefficients

Source	Estimate	Standard Error	T-Statistic	p-value
Comp.1 – (Pyrolytic)	66,631	2,20512	30,2164	0,0000
Comp.2 – (Petrogenic)	40,3759	2,49963	16,1527	0,0000
Comp.3 – (Diagenetic)	19,5241	7,78436	2,5081	0,0334
Constant	266,884	6,50584	41,0222	0,0000

into the river. Additionally, besides these inputs, some other sources such as the roads on both sides of the river and along the coast, urban runoff containing street dust and municipal wastewater resulted in the pattern of pyrolytic origins of PAH contamination in the area.

5. CONCLUSIONS

This study was the first to document a comprehensive analysis of PAHs in the Sarno River and Estuary, and has provided very useful information for the evaluation of trace PAHs and probable sources of PAHs. Source analysis revealed that PAHs mainly came from combustion processes. PCA indicated that PAH contamination in the Sarno River and Estuary resulted from a mixed pattern of pyrolytic and petrogenic sources; however, a predominant pattern of pyrolytic inputs of PAHs was found. Particularly, the total variance in the composition of PAHs was 51.2%, 39.9% and 4.1% from pyrolytic, petrogenic and diagenic sources, respectively. The ratio study reflected a prevailing pattern of pyrolytic inputs of PAHs in sediments from the Sarno River and Estuary. In this study, PCR was a very good statistical technique for handling the problem of multicollinearity by eliminating model instability and reducing the variances of the regression coefficients. The PCR coefficients for the pyrolytic, petrogenic and diagenic components were 66.6, 40.4 and 19.5, respectively. Thus, the mean percentage contribution to total PAH levels were 52.6% from the combustion of coal and wood, 31.9% from oil-derived sources and 15.43% from natural sources.

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